

Biology More Basic Chemistry Practice: Name: _____ Hr: _____

Remember: (protons=electrons=atomic Number): Neutrons=(atomic mass rounded to a whole number) minus- (atomic number or number of protons)

Complete---fill in appropriate number for each element.

	<u>Protons</u>	<u>Electrons</u>	<u>Neutrons</u>	<u>Atomic Mass</u>
<u>Phosphorous</u>	_____	_____	_____	_____
<u>Titanium</u>	_____	_____	_____	_____
<u>Magnesium</u>	_____	_____	_____	_____
<u>Fluorine</u>	_____	_____	_____	_____
<u>Platinum</u>	_____	_____	_____	_____
<u>Copper</u>	_____	_____	_____	_____
<u>Iron</u>	_____	_____	_____	_____

Complete---Draw the nucleus and electron shells. Label and fill in the number of protons, neutrons, and electrons. Remember the 1st level can hold 2 max, 2nd 8 max, and 3rd 18 max.

1. Nitrogen

2. Silicon

3. Beryllium

4. Argon

5. Sulfur

6. Chlorine

7. Nickel